

# ADCI Supervisor's Certification Examination

## Diving Formula Worksheet

### PSIG to PSIA

$$\text{PSIA} = \text{PSIG} + 14.7$$

Round up to the next whole number

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### PSIA to PSIG

$$\text{PSIG} = \text{PSIA} - 14.7$$

Round up to the next whole number

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### Depth (fsw) to PSIG

$$\text{PSIG} = \text{Depth} \times .445$$

Round up to next whole number

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### PSIG to Depth (fsw)

$$\text{Depth} = \text{PSIG} \text{ divided by } .445$$

Round up to next whole number

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### PSIG to Atmospheres Absolute (ATA)

$$\text{ATA} = \frac{(\text{PSIG} + 14.7)}{14.7}$$

Carry two decimal places

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### Atmospheres Absolute (ATA) to PSIG

$$(\text{ATA} - 1) \times 14.7 = \text{PSIG}$$

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### Depth (fsw) to Atmospheres Absolute (ATA)

$$\text{ATA} = \frac{\text{Depth} + 33}{33}$$

Carry two decimal places

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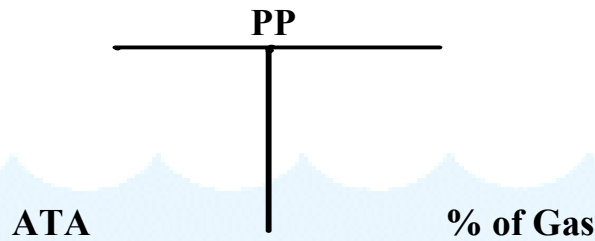
### ATA to Depth (fsw)

$$\text{ATA} - 1 \times 33 = \text{Depth (fsw)}$$

Round up to next whole number

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## DALTON'S LAW ("T" Formula)



PP = Partial Pressure  
% = Percent by Volume of the identified gas  
ATA = Atmospheres Absolute

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## GAY-LUSSAC'S LAW

$$P2 = \frac{P1 \times T2}{T1}$$

Volume is constant  
T1 = Initial Temperature (absolute)  
T2 = Final Temperature (absolute)  
P1 = Initial Pressure (absolute)  
P2 = Final Pressure (absolute)

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## CHARLES' LAW

$$V2 = \frac{P1 \times T2}{T1}$$

Pressure is constant  
T1 = Initial Temperature (absolute)  
T2 = Final Temperature (absolute)  
P1 = Initial Pressure (absolute)  
P2 = Final Pressure (absolute)

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## Boyles Law (Pressure/Volume Relationship)

$$\frac{DL + 33}{DA + 33} \times OV = NV$$

DL = Depth Left  
DA = Depth Arrived  
OV = Original Volume  
NV = New Volume

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## Henry's Law

### (The Law of Gas Absorption and Solubility) EXPLANATION:

- “The amount of any given gas that will dissolve in a liquid at a given temperature is directly proportional to the partial pressure of that gas.”
- Gas diffuses and dissolves in blood, because of the difference in partial pressure, between inhaled and exhaled air.
- The inert gas in the breathing media, (nitrogen or helium), will be dissolved into the diver's body tissues as he is descending and during the time spent on bottom.
- Whatever gases that has been dissolved in a diver's body tissues, at a given depth and pressure, will remain in the tissues, as long as the depth is maintained. As the diver starts to ascend, more and more of the dissolved gas will come out of his tissues. If his ascent is controlled, as through the use of the decompression table, the dissolved gas will be carried to the lungs and exhaled, before it accumulates sufficiently to form significant bubbles in the blood or tissues.

### General Gas Law (Pressure/Volume/Temperature Relationship)

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$$(P1 \times V1) \div T1 = (P2 \times V2) \div T2$$

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#### Degrees Fahrenheit to Rankine

$$R^{\circ} = F^{\circ} + 460^{\circ}$$

#### Degrees Celsius to Absolute

$$C^{\circ} + 273^{\circ} = \text{Degrees Kelvin}$$

#### Degrees Fahrenheit to Celsius

$$5 \times (F^{\circ} - 32^{\circ}) \div 9 = \text{Celsius (carry 1 decimal place)}$$

#### Degrees Celsius to Fahrenheit

$$(9 \times C^{\circ}) \div 5 = 32^{\circ} = \text{Fahrenheit (carry 1 decimal place)}$$

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### Gas Volume Requirement Formula If Using an LP Compressor

$$SCFM = ATA \times ACFM \times N$$

### Gas Volume Requirement Formula If Using an HP Gas Bank

$$SCF = ATA \times ACFM \times N \times T$$

SCFM = Standard Cubic Feet per Minute

SCF = Standard Cubic Feet

ATA = Atmospheres Absolute

ACFM = Actual Cubic Feet Per Minute

N = Number of Divers

T = Time (Always expressed in minutes)

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### Minimum Manifold Pressure

**MMP = D x .445 + Over Bottom Pressure**  
**(OBP is established by company or a set standard used)**  
**Round up to next whole number**

### Gas Consumption

<b>Free-Flow Type Hat(Desco, MK V)</b>	<b>4.5 ACFM</b>
<b>Demand Type Hat(Superlite/Miller)</b>	<b>1.4 ACFM</b>
<b>Built-in-Breathing System(BIBS)</b>	<b>0.3 ACFM</b>

### Treatment Gas Mixtures ( O<sub>2</sub>/HeO<sub>2</sub>/N<sub>2</sub>O<sub>2</sub> )

<b>Depth (fsw)</b>	<b>Gas Mixture</b>	<b>PPO<sub>2</sub></b>
<b>0 – 60 fsw</b>	<b>100 % O<sub>2</sub></b>	<b>1.00 – 2.81 ATA</b>
<b>61 – 165 fsw</b>	<b>50/50% HeO<sub>2</sub> or N<sub>2</sub>O<sub>2</sub></b>	<b>1.42 – 3.00 ATA</b>
<b>166 – 225 fsw</b>	<b>64/36% HeO<sub>2</sub></b>	<b>2.17 – 2.80 ATA</b>

### Calculating In-Water Travel Time

- 1. Depth left (ft/m) minus(-) Depth arrived (ft/m) = Distance Traveled (ft/m)**
- 2. Distance Traveled divided(÷) by ascent/descent rate = minutes (and/or percentage of a minute in decimal**
- 3. Whole number is minute(s). Decimal is percentage of minute. Take decimal and multiply(x) by 60 (number of seconds in a minute). Decimal will then convert to actual seconds. FOR EXAMPLE:**

215 fsw – 87 fsw = 128 fsw      Ascent rate: 30 fpm  
128 fsw ÷ 30 fpm = 4.26 ( 4 minutes and .26 or 26% of a minute)  
.26 x 60 = 15.6 seconds (round up to next whole second)= 16 seconds  
:4::16 is your travel time from 215' to 87'

### One Example of: Calculating Surface Interval

Reached surface (RS) @ 2305 hrs.  
Left surface (LS) @ 0317 hrs. ( carry over 24 hr. clock )  
0317 hrs. could be expressed, ONLY FOR THE PURPOSE OF CALCULATION, as 2717 hrs. 2717 minus(-) 2305 = 4:12  
4 hrs. and 12 min.

# Salvage Formulas

## Area of a Square or Rectangle

Length x Width

## Area of a Cylinder

Ends =  $(\frac{1}{4}\pi \times D^2) \times 2$

Tube =  $(\pi \times D) \times L$

Ends + Tube = Total Area

## Volume of a Cube

Length x Width x Height = Volume

## Volume of a Cylinder

$(\pi \times D^2) \div 4 \times \text{Height} = \text{Volume}$

## Lifting Capacity

Volume x Weight of Water – Weight of Lifting Device = Lifting Capacity